

Module 3

H₂ molecular emission and collisional-radiative modelling

Lecture viewgraphs

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ADAS/ADAS-EU

- 1. Preliminaries and nomenclatures.
- 2. The molecular H2 database.
- 3. Vibronic collisional-radiative modelling.
- 4. Rovibrational spectral simulation and comparison.
- 5. Conclusions.

1.1 Preliminaries



ADAS has restricted interest to diatomic molecules only.

There are two primary motivations:

- (a) analysis of observations of band spectral features at visible wavelengths, which discriminate isotopomers, temperatures $(T_r \& T_y)$ and clarify sources.
- (b) Collisional-radiative modelling of H₂ isotopomers for plasma models and quantitative spectroscopy in the visible and UV.

For the former, ADAS models rovibrational band wavelengths between electronic states, with assumptions of upper emitting state populations.

For the latter, ADAS has developed a vibronic collisional radiative model for the H₂ system. This model can provide upper state population information for quantitative/calibrated spectroscopy of hydrogen molecular emission.

Following the ADAS practice for atoms and ions, a number of data formats have been introduced for the molecular problem. They are called *molecular data formats* (such as *mdf00*). These fall into the two categories of fundamental data and derived (collisional-radiative) data.

We have tried as far as possible to follow the data format numbering of the atomic case. Thus *mdf00*, *mdf01*, *mdf02* and *mdf04* are fundamental data collections. *mdf04* is a *specific molecular system file* which, as for its atomic counterpart, can support a collisional-radiative (vibronic) model.

In magnetic confinement fusion research, it is necessary to work with various isotopic combinations, such as H_2 , D_2 , T_2 , HD, HT and DT. It is assumed that the electronic states and their electronic transitions are independent of the nuclear masses. The vibrational and rotational sub-states do reflect the specific isotopic masses.

In practical terms, molecular studies must focus on molecular systems. Thus the D_2 system includes D_2 , D_2^+ , D, D⁺ (and in principle D_3 and D_3^+) and these must be treated together in collisional - radiative modelling. For example, the direct formation of D(n=3) by molecular dissociation is an important parameter.

Molecular constants:

- mdf00: enu sub-directories by isotopomer separate datasets of vibrational substate energies for each electronic state.
 - pot separate datasets of electronic potential curve for each electronic state.
 - dip separate datasets of electronic dipole curve for each electronic state.
 - aval sub-directories by isotopomer separate datasets of vibrationally resolved A-values for each electronic state pair.
 - fcf sub-directories by isotopomer separate datasets of vibrationally resolved Franck-Condon factors for each electronic state pair.
- mdf01: nist sub-directories by isotopomer extract of NIST molecular constant table for selected electronic states.
 - behr sub-directories by isotopomer driver parameter file for ADAS908 rovibrational spectral analysis - relevant electronic and vibrational transitions.
 - pickett sub-directories by isotopomer driver parameter file for ADAS908 rovibrational spectral analysis - relevant electronic and vibrational transitions.

2.1 Electron impact processes with the H₂ molecule

The electron collision processes with the H₂ molecular system has been studied for many years and compilations of reaction cross-sections assembled. A major revision has taken place by Janev, Reiter and Guzman linked to the ADAS-EU project to put these data in a form suitable for ADAS collisional-radiative modelling.

Fundamental data in ADAS data formats are archived as preferred numerical tabulations, with specified asymptotic behaviour. In the molecular collision area, algebraic forms are often used (in certain cases multiple forms) with some lack of clarity in preferment. The ADAS database seeks to correct this.

The ADAS impact process molecular data format is *mdf02* and it has been designed to have both numerical values and the parameters of the formulaic representation (if available) of each transition.

The primary dataset for electron impact with H₂ is *mdf02/fg13_h2#e.dat*. The groups of processes are summarised below.

Excitation:	$e + H_2^{(+q)}(N,\nu) \rightarrow e + H_2^{(+q)}(N',\nu')$ vibrationally resolved
	$e + H_2(N) \rightarrow e + H_2(N) \qquad \text{vibrationally unresolved}$
	$e + H_2(N,\nu) \rightarrow e + H_2(N',\nu^{**}) \rightarrow e + H(1s) + H(nl) \text{ dissociative}$
	$e + H_2(X, \nu) \to H_2^-(X^2\Sigma_u^+, B^2\Sigma_u^+) \to e + H_2(X, \nu') \text{vibrational via e-attachment}$
lonisation:	$e + H_2(N, \nu) \rightarrow e + e + H_2^+(N', \nu') \text{vibrationally resolved}$
	$e + H_2(N) \rightarrow e + e + H_2^+(N')$ vibrationally unresolved
	$e + H_2(N,\nu) \rightarrow e + e + \tilde{H}_2^+(N',\nu'^{**}) \rightarrow e + H + H^+$ dissociative
	$H_2(N^{**}, \nu^{**}) \rightarrow H_2^+(N', \nu') + e + h\nu \qquad \text{autoionisation}$

2.2 Electron impact processes with the H₂ molecule (contd.)

Dissociation: (excluding above)

$$\begin{split} &e + H_2^+(N,\nu) \to e + H_2^+(d) \to e + H(1s) + H(nl) & \text{vibrationally resolved} \\ &e + H_2^+(N) \to e + H_2^+(d) \to e + H(1s) + H(nl) & \text{vibrationally unresolved} \\ &e + H_2(N,\nu) \to H_2^-(X\,^2\Sigma_u^+) \to H^- + H(1s) & \text{diss. er attachment via } H_2^-(X) \text{ and } H^- \text{ production} \\ &e + H_2(N,\nu) \to H_2^-(B\,^2\Sigma_g^+) \to H^- + H(1s) & \text{diss. er attachment via } H_2^-(B) \text{ and } H^- \text{ production} \\ &e + H_2(X,\nu) \to H_2^-(X/B^2\Sigma_{(u/g)}^+) \to e + H(1s) + H(1s) & \text{diss. er attachment} \end{split}$$

$$\begin{array}{ll} \mbox{Recombination:} & H + H_2^+(N,\nu) \to H^+ + H_2(N',\nu') & \mbox{vibrationally resolved inverse CX} \\ & H + H_2^+(N) \to H^+ + H_2(N) & \mbox{vibrationally unresolved inverse CX} \\ & e + H_2^+ \to H_2^{**} \to H(n) + H(n) & \mbox{dissociative recombination} \end{array}$$

Charge exchange:

 $\begin{array}{ll} \mbox{ge:} & H^+ + H_2(N,\nu) \rightarrow H(1s) + H_2^+(N',\nu') & \mbox{vibrationally resolved direct CX} \\ & H^+ + H_2(N) \rightarrow H(1s) + H_2^+(N') & \mbox{vibrationally unresolved direct CX} \\ & H^+ + H_2(N,\nu) \rightarrow H(1s) + H + H^+ & \mbox{dissociative CX} \\ & H^+ + H_2(\nu) \rightarrow H(1s) + H^+ + H^+ + e & \mbox{transfer ionisation} \end{array}$

There is a corresponding dataset for H⁺ impact processes with H₂ called *mdf02/fg13_h2#p.dat*

2.3 H₂ system excitation data of format *mdf02*

system	→ H_2 /	/ electro	on impact				md	£02	
species list with characterisations	species / / ind_s / 1 2 3 4 5 6	identity H_2 H_2+ H_2^+ H_2^- H H^++ H^+	e-config dia-ua dia-ua dia-ua dia-ua atm atm atm	dia-to dia-to dia-to dia-to dia-to dia-to atm-to atm-to	ling ch_ion erm 1->2 erm 2->5+5 erm 3->1 erm 4->5 erm N erm 6->4	bwno_i 124418. 124418. 5386. 0. 109691. 0. 427.	ch_dis 1->4+4 1->5+6 2->4+5 3->4+6 N N N	bwno_d 36117.00 36117.00 21380.21 00000.00 00000.00 00000.00 00000.00	
process list with descriptions	/ process / / //ind_p / 1 11 53 54 55 / / states	$path$ $e+H_2(^+q)(N,v) \rightarrow e + H_2(+q)*(N',v')$ $e+H_2(N,v) \rightarrow e + H_2*(N',v')$ $H^+ + H_2(N,v) \rightarrow H(1s) + H + H^+$ $H^+ + H_2(v) \rightarrow H(1s) + H^+ + H^+ + e$				exci vib diss tran othe	description tation resolved exc iociative CX sfer ion. r CX process	 • es	
	/ ind_s	ind_e	e-con-ua	e-con-sa	coupled state	e (wt1)/2	wno	wno_d	comments
	, 1 1	1 1 2 1	ssg1ssg ssg2psu	1s1s 1s2p	(1)S(+)(g) (1)S(+)(u)	0.0 0.0	0.0 91691.69	38292.98 120595.97	ground state X B
electronic level list	> 1 2 2 2 4 5 /	29 1 1 1 2 2 3 2 1 1	lssg4dpg lssg 2psu 2ppu	1s3d 1s 2s 1s	(3) P()(g) (2) S(+)(g) (2) S(+)(u) (2) P()(u) (2) P()(u) (2) S	2.5 0.5 0.5 0.5 0.5 0.0	118509.87 124418.00 274666.12 318289.02 148030.30	135837.26 148030.30 274666.12 318289.02 148030.30	r dissociative state dissociative state ion H^+
electronic level list	<pre>/ states / / ind_s / 1 1</pre>	ind_e 1 1 2 1 29 1 1 1 2 2 3 2 1 1	e-con-ua .ssg1ssg .ssg2psu .ssg4dpg .ssg 2psu 2ppu	e-con-sa 1s1s 1s2p 1s3d 1s 2s 1s	<pre>coupled state (1) S (+) (g) (1) S (+) (u) (3) P() (g) (2) S (+) (g) (2) S (+) (g) (2) S (+) (u) (2) P() (u) (2) S</pre>	e (wt1)/2 0.0 0.0 2.5 0.5 0.5 0.5 0.5 0.5 0.0	wno 0.0 91691.69 118509.87 124418.00 274666.12 318289.02 148030.30	wno_d 38292.98 120595.97 135837.26 148030.30 274666.12 318289.02 148030.30	comments ground state X B r dissociative state dissociative state ion H^+

2.4 Transition data and the ADAS901 processing screen



2.5 Text and graphical output from ADAS901



The ADAS codes ADAS903 and ADAS904 operate in sequence. From *mdf02*, ADAS903 converts cross-sections to rate coefficients. It prepares an intermediate dataset of format *mdf33* structurally similar to *mdf02*.

ADAS904 gathers the level list and reaction data from *mdf33* and includes A-values and vibrational energies from *mdf00* to prepare the vibrationally-resolved specific molecule file *mdf04*.

Formally, *mdf04* is similar to *adf04*. It is sufficient to support a vibronic population calculation.

ADAS905 executes the vibronic collisional-radiative model for the population solution. It generates the generalised collisional-radiative coefficients of format *mdf11* which couple metastable states of species together and the associated emissivity coefficients *mdf15*. These names parallel the *adf11* and *adf15* formats for atoms and ions.



1 3 7

The generalised collisional matrix reductions follow broadly the same path as for atoms and ions. But there are some new coefficients

The prefix \mathcal{M} is used for the molecular generalised collisional-radiative coefficients.

New inverse charge exchange , recombination coefficient

New charge exchange ionisation coefficient

New dissociative coefficients. —

$$\frac{dN_{\rho}}{dt} = \mathcal{M}C_{\rho\sigma}N_{\sigma} + \mathcal{M}Q_{\rho\nu}^{ICX}N_{\nu}N_{H} + \mathcal{M}\mathcal{R}_{\rho\nu}N_{\nu}N_{H}$$

$$\mathcal{M}C_{\rho\sigma} = \begin{bmatrix} C_{\rho\sigma} - C_{\rho j}C_{ji}^{-1}C_{i\sigma} \end{bmatrix} / N_{e} \qquad \mathcal{M}2\mathcal{C}\mathcal{D}$$

$$\mathcal{M}\mathcal{R}_{\rho\nu} = r_{\rho\nu} - C_{\rho j}C_{ji}^{-1}r_{i\nu} \qquad \mathcal{M}\mathcal{A}\mathcal{C}\mathcal{D}$$

$$\mathcal{M}Q_{\rho\nu}^{ICX} = Q_{\rho\nu}^{ICX} - C_{\rho j}C_{ji}^{-1}Q_{i\nu}^{ICX} \qquad \mathcal{M}\mathcal{C}\mathcal{X}\mathcal{F}\mathcal{A}\mathcal{C}\mathcal{D}$$

$$\mathcal{M}S_{\nu\sigma} = S_{\nu\sigma} - S_{\nu j}C_{ji}^{-1}C_{i\sigma} \qquad \mathcal{M}\mathcal{F}\mathcal{C}\mathcal{D}$$

$$\mathcal{M}Q_{\nu\sigma}^{CX} = Q_{\nu\sigma}^{CX} - Q_{\nu j}^{CX}C_{ji}^{-1}C_{i\sigma} \qquad \mathcal{M}\mathcal{C}\mathcal{X}\mathcal{D}$$

$$\mathcal{P}\mathcal{D}_{\mu\sigma} = \mathcal{D}_{\mu\sigma} - C_{\mu j''}C_{j''i''}\mathcal{D}_{i''\sigma} \qquad \mathcal{P}\mathcal{D}\mathcal{C}\mathcal{D}$$

$$\mathcal{P}\mathcal{A}\mathcal{D}_{\mu\nu} = -\mathcal{D}_{\mu i}C_{ii}^{-1}Q_{i\nu}^{ICX}N_{H} - \mathcal{D}_{\mu i}C_{ii}^{-1}r_{i\nu}N_{e} \qquad \mathcal{P}\mathcal{A}\mathcal{D}\mathcal{C}\mathcal{D}$$

$$\mathcal{P}\mathcal{A}\mathcal{D}\mathcal{S}_{\mu^{+}\nu} = -N_{e}S_{\mu^{+}j''}C_{j''i''}\mathcal{D}_{i''}C_{ji}^{-1}\left[Q_{i\nu}^{ICX}N_{H} + r_{i\nu}N_{e}\right] \qquad \mathcal{P}\mathcal{A}\mathcal{D}\mathcal{F}\mathcal{D}$$

3.3 H₂ ground state vibrational population structure

In the H2 system, all the vibrational states of the ground electronic state H_2 (X ${}^1S_g^+$) are metastable, since they cannot radiate to the lowest vibrational state.

The figures to the right show the vibrational substate populations of the ground state in equilibrium.

The vibrational states of the excited electronic states are not metastable. They are ordinary states in the generalised collisional-radiative description.

The set of generalised collisional-radiative coefficients is therefore much larger for the molecular case. This can be a bit overwhelming for basic studies, so the coefficients are often summed and averaged over the vibrational substates of the ground electronic state.



3.3 H₂ collisional-radiative coefficients

The coefficients shown here have been summed over the ground vibrational substates. In the most consistent practice the separate vibrational sub-states of the H_2 (X ${}^1S_g^+$) ground electronic state are all metastable and should be explicit.

Dissociation coefficient \mathcal{PDCD} for the effective dissociation pathway: $H_2 (X \ ^1\Sigma_g^+) \rightarrow H + H(H^+)$





lonisation coefficient *MSCD* for the pathway : H₂ (X ¹Σ_g⁺) → H₂⁺ (X ²Σ_g⁺) +e

4.1 Spectral analysis using ADAS908

Practical spectral analysis for complex molecular rovibrational spectra often proceeds by comparison and matching of theoretical and experimental spectra over wavelength intervals.

If a convincing match can be obtained, then the variation of relative intensities as a function of effective rotational and vibrational temperatures can be exploited.

ADAS908 is provided to assist in such investigation. It can work with several spectra (experimental and theoretical) simultaneously. Control values appropriate to the experimental spectra and to the theoretical spectra and to the display can be adjusted giving substantial flexibility.



The usual parametric expansion of a rovibrational level energy is written as:

$$T_{\nu,J} = G(\nu) + F_{\nu}(J)$$

= $\omega_e(\nu + 1/2) - \omega_e x_e(\nu + 1/2)^2 + \omega_e y_e(\nu + 1/2)^3 + \dots$
+ $B_{\nu}J(J+1) - D_{\nu}J^2(J+1)^2 + H_{\nu}J^3(J+1)^3 + \dots$

where the rotational parameters may have the supplementary expansions:

$$B_{\nu} = B_{e} - \alpha_{e}(\nu + 1/2) + \beta_{e}(\nu + 1/2)^{2} + \dots$$

$$D_{\nu} = D_{e} - \gamma_{e}(\nu + 1/2) + \dots$$

If there is no upper electronic state vibrational sub-population collisional-radiative model, then upward projection of the ground and/or metastable thermal vibrational distributions is assumed according to Franck-Condon factors. This can be handled by adjusting B_{ν} values and populations as

$$\bar{B}'_{k,\nu'} = \left[\sum_{\nu_{g}} B_{g,\nu_{g}} F^{k,\nu'}_{g\nu_{g}} N_{g,\nu_{g}} + \sum_{\nu_{m}} B_{m,\nu_{m}} F^{k,\nu'}_{m\nu_{m}} N_{m,\nu_{m}}\right] / \left[\sum_{\nu_{g}} N_{g,\nu_{g}} + \sum_{\nu_{m}} N_{m,\nu_{m}}\right]$$
$$N_{k,\nu',J} = \omega_{J} exp(-\bar{B}'_{k,\nu''} J(J+1)/kT_{r}) N_{k,\nu'} / \sum_{J} \omega_{J} exp(-\bar{B}'_{k,\nu'} J(J+1)/kT_{r})$$

which most conveniently is presented as a modified rotational temperature $T'_r = B'_{k,\nu'}T_r/\bar{B}'_{k,\nu'}$

The *mdf01/behr* drivers have been specially tuned to the H_2 molecular case.

The nuclear spin specification is required for homonuclear systems so that the intensity alternation of successive rotational transitions may be correctly determined.

Specific rotational parameters for vibrational substates support the high wavelength precision needed for H2 rovibrational identification.

> +++ -1

HH; mo KS1	df01 Behringer d = 0.5	format;	etran:	s = all; vtra	ans = all		
KS2	= 0.5	nuclear sp	oins				
1 2 3 4 -1	X singlet Sigma B singlet Sigma a triplet Sigma d triplet Pi	ag au ag u		0.000 91700.000 95936.1 112700.3	electronic states		
1	e=1 v= 0		1	2179.586	59.3634	0.0455	0.0000465
15 16	e=1 v=14 e=2 v= 0		1 2	38154.391 682.321	8.7701 19.4563	0.08467 0.0156	0.0 0.0000163
30 31 32 -1 1 2 -1	$ \begin{array}{c} e=2 v=14 \\ e=3 v=* \\ e=4 v=* \\ 2 1 \\ 4 3 \\ \end{array} $	2 16139.337 3 0. 4 0. vite cor spectral generation			11.375 vibrational states	0.0035 improved param specific st	0.0000127 d rotational eters for vibronic ates
state vmax jmax om omx omy Be De alp bet He Te	<pre>= "X" = 14 = 20 = 4401.2654 = 120.602 = 0.7242 = 60.8477 = 0.04653 = 3.0513 = 0.0 = 5.168E-05 = 0.0</pre>	e	base ro param ach ele	ovibrational eter set for ectronic state			

4.4 The mdf01/pickett driver for BeD

par

The method is best suited to diatomics with heavier nuclei than H₂ for which vibrational and rotational energy separation is better.

Examination and selection of parameters was carried out by Duxbury for JET experiments. ADAS has such special data for CH, CD, C₂, BeH, BeD and BeT.

The capabilities of the Pickett package are very large, but expert knowledge is required for full utilisation. The ADAS mdf01/pickett database will be extended as needed.



BeD; mdf01 Pickett format; etrans = 2 Sigma - 2 Pi doublet Sigma 0.000 electronic states 2 doublet Pi 20037.910 -1 e=1 v=0 1 0.000 8292.458 vib. substates 1 e=1 v=6 e=2 v=0 2 0.000 e=2 v=6 2 8320.162 14 -1 1 8 1 vibronic pairs with 9 2 2 detailed parameters 3 3 10 11 following 4 -1 1 beryllium deuteride 2pi-2sig 14-Oct-2003 Pickett 1010 79002 494.139 0 120 -45.0 -45.0 600000 TTTTT int 013 0.0100 data 14 Oct 2003 Beryllium deuteride 0-0 band 2pi-2sigma 16 600 3 1.1220000E-10 1000.000 1.000 'a' -2 2 0 0 0 2 -1 1 1 1 -2 1 1 1 0 2 1 1 0 1 11 20037.68838 .05 ! E' (Band Centre) 9.59E - 8100 5.62479800 ! Bbar" (1sig) 111 5.69574400 9.59E-8 ! Bbar' (1pi) Pickett Pickett 200 -3.1176800E-4 2.00E-10 ! DJ" (1siq) datablock 211 -3.1670100E-4 2.00E-10 ! DJ' (1pi) 2.00E-10 300 1.6238000E-8 ! HJ" lsig) for each data 311 1.5743000E-8 2.00E-10 ! HJ' (1pi) vibronic 400 -0.8746000E-12 2.00E-10 ! LJ" 1sig) 411 -0.9078000E-12 2.00E-10 ! LJ' (1pi) 40011 2.1096000e-3 2.00e-10 1 q/2 lambda 40111 -4.5860000e-7 2.00e-10 dq/2 lambda 40211 2.6995000e-11 2.00e-10 hg/2 lambda 10010011 2.1592000 2.00e-10 Aso 2pi 1000011 -0.801500e-2 2.00e-10 gamma 2pi 1000111 0.806900e-5 2.00e-10 dgamma 2pi 1000211 -2.218000e-9 2.00e-10 hgamma 2pi parameter parameter -1 codes values

pair

4.5 ADAS908 plot and analyse window

The processing w ADAS908 adjusts to the number of being handled.

Refreshing, re-eva spectra by repixel model re-computa according the par modified.

	ADAS908 GRAPIC PROCESSING	
ng window of justs according er of spectra d. e-evaluates the pixellation or by nputation e parameters		
Selected theoretical spectra from model calculation	Display controls xmin/nm xmax/nm npix ymin ymax 598,000 608,000 1000 0.00000 1.00000 refresh print Theoretical spectrum Graph controls multiplier width/nm colour T_vib/K T_rot/K v_min v_max J_min J_max mdf01/behr/h2/kub12_h2_a-d.dati 0.03000 0.30000 1.00000 0.0100000 1 3000.00 100 1 10 mdf01/behr/h2/kub12_h2_a-d.dati 0.00000 0.60000 1.00000 0.0100000 1 3000.00 100 10 1 10	
selected experimental spectra	Experimental spectrum anme reh908/d2/kub12_d2_fulcher.dat Cancel Done X-y shifting, scaling, resolutions and colour resolutions and colour parameters	est
	manipulation	

- A revision and update of the reaction process for the H₂ molecular system has been carried out and assembled in a new ADAS molecular data format *mdf02*.
- Supplementary molecular vibrational energy levels, A-values, Franck-Condon factors and other fundamental data have been assembled in molecular data format *mdf00*.
- A code machinery has been set up in ADAS series ADAS9xx to prepare the specific molecule dataset of format *mdf04*.
- A generalised collisional-radiative model of the H₂ system in vibronic resolution has been implemented called ADAS905.
- The rovibrational relative spectral prediction of diatomic band structure has been set based on two models the *Behringer model* and the *Pickett model* suited to the H₂ system and non-homogenous diatomics respectively. These use driver datasets of molecular data format *mdf01*.
- A code ADAS908 has been set up for comparative assessment of experimental diatomic spectra and theoretical spectra created with *mdf01/behr* and *mdf01/pickett* drivers.